QP CODE:104018	Reg. No:
First Year B.Sc (MRT) Degree So	upplementary Examinations March 2018
Atomic and Nuclear Physics	
Time: 3 Hours	Total Marks: 100
<ul><li>Answer all Quest</li><li>Draw Diagrams w</li></ul>	ions. herever necessary. (2x20=40)
Describe the construction and working	g of nuclear reactor
2. Explain the production, properties of cathode rays in Thomson's model. Interpret the e/m ratio of electron by Thomson experiment.	
Short notes:	(8x5=40)
3. Explain about natural and artificial rad	ioactivity.
4. Discuss Somerfield's atom model.	
5. Derive an expression for an half life of radioactive substance in terms of decay.	
Mention the relation between mean lif	e and decay constant. What is the SI unit of
6. State four distinguish properties of β a	and Y.
7. Explain the limitation of Bohr's atom model.	
8. What is $\beta$ decay. Explain the various $\beta$	process involved in β decay.
9. Distinguish between isotope and isob	ar with an example. What is isotone.
10. State the laws of photoelectric emission	
Answer briefly:	(10x2=20)
11. Define radioisotope with an example.	
12. State the properties of neutrons.	
13. Express 16mg mass into equivalent energy in electron volt	
14. What do you mean by Q value of a nuclear reaction	
15. State the postulates of Bohr's theory of hydrogen atom.	
16 Define the appoint abarge of electron, what is the value of e/m of eathede rave	

- 16. Define the specific charge of electron, what is the value of e/m of cathode rays.
- 17. Define excitation and ionization potential.
- 18. What are different types of mesons
- 19. What is Zeemen effect
- 20. State De-Broglie's theory.

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